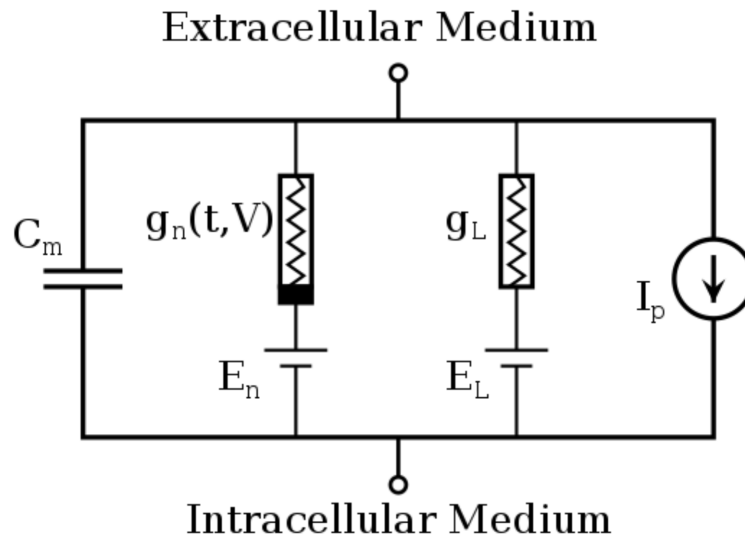


# Outline

- HH Model Revisited
- Nonlinear Dynamics in HH Model

## HH Model Revisited

### Introduction



(figure from Internet)

In HH model, we use 3 additional ODEs to replace the 2 imposed eqs of IF model. And its first ODE is also gotten from its circuit diagram.

$$I = C_m \frac{dV_m}{dt} + \bar{g}_K n^4 (V_m - V_K) + \bar{g}_{Na} m^3 h (V_m - V_{Na}) + \bar{g}_l (V_m - V_l)$$

$$\frac{dn}{dt} = \alpha_n(V_m)(1 - n) - \beta_n(V_m)n$$

$$\frac{dm}{dt} = \alpha_m(V_m)(1 - m) - \beta_m(V_m)m$$

$$\frac{dh}{dt} = \alpha_h(V_m)(1 - h) - \beta_h(V_m)h$$

(figure from Internet)

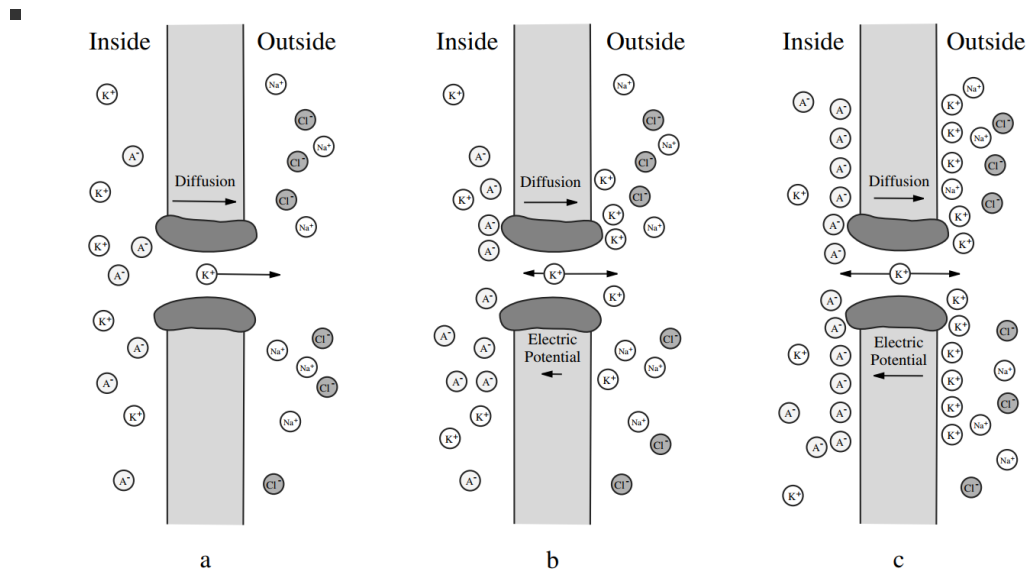
### Physical Perspective

- What is the meaning of  $\alpha$  and  $\beta$ ?
  - Transition probability in the 2 state Markov chain.
- Why the exponent is 4, 3, 1 for  $n, m, h$  in the first ODE?
  - They come from fitting the experiment data.

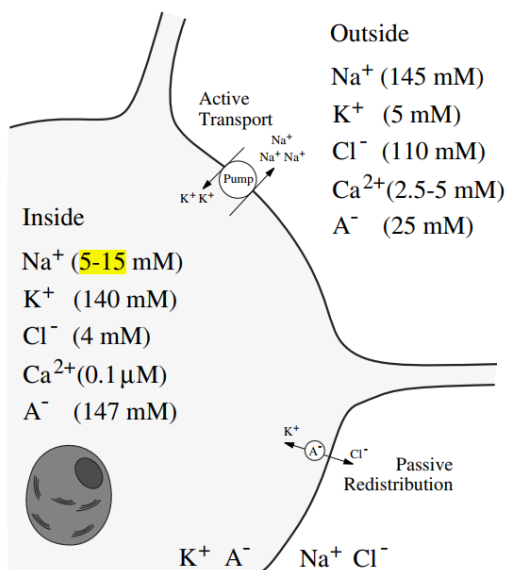
# Biological Perspective

There are a few things you need to know:

- How to calculate the  $V_K$  and  $V_{Na}$ ? (Sometimes they are called Nernst Equilibrium Potential)
  - Method 1: using diffusion equation



- Method 2: using Boltzmann distribution
- What is the quantity of Nernst Equilibrium Potential of typical ions?



## Equilibrium Potentials

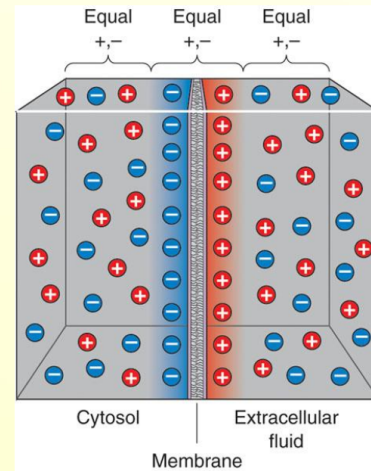
$$\begin{aligned}
 \text{Na}^+ & 62 \log \frac{145}{5} = 90 \text{ mV} \\
 & 62 \log \frac{145}{15} = 61 \text{ mV} \\
 \text{K}^+ & 62 \log \frac{5}{140} = -90 \text{ mV} \\
 \text{Cl}^- & -62 \log \frac{110}{4} = -89 \text{ mV} \\
 \text{Ca}^{2+} & 31 \log \frac{2.5}{10^{-4}} = 136 \text{ mV} \\
 & 31 \log \frac{5}{10^{-4}} = 146 \text{ mV}
 \end{aligned}$$

Ion	Concentration outside (in mM)	Concentration inside (in mM)	Ratio Out : In	E <sub>ion</sub> (at 37°C)
K <sup>+</sup>	5	100	1 : 20	-80 mV
Na <sup>+</sup>	150	15	10 : 1	62 mV
Ca <sup>2+</sup>	2	0.0002	10,000 : 1	123 mV
Cl <sup>-</sup>	150	13	11.5 : 1	-65 mV

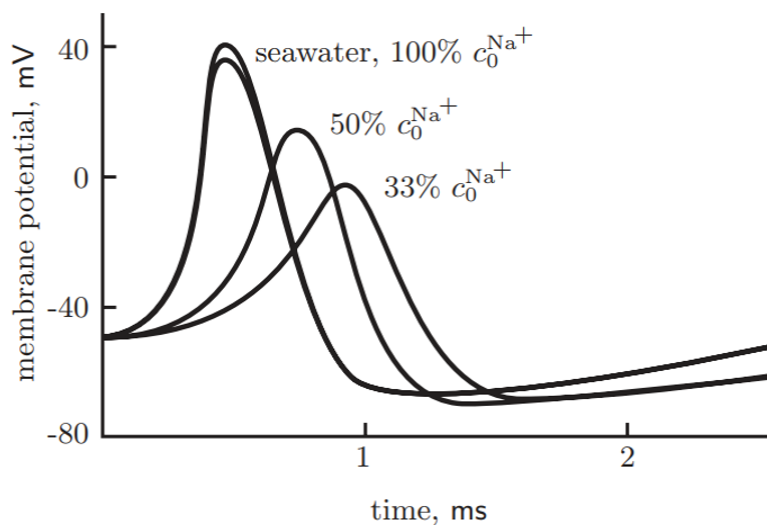
- What are the concentration of  $Na^+$ ,  $K^+$ ,  $Cl^-$ ,  $Ca^{2+}$  inside the neuron and outside the neuron?
  - see above.
- How much will they change during the action potential?
  - $K^+$

For a cell with a  $50\text{ }\mu\text{m}$  diameter, containing  $100\text{ mM } K^+$ , it can be calculated that the concentration change required to take the membrane from  $0$  to  $-80\text{ mV}$  is about  **$0.00001\text{ mM}$** .

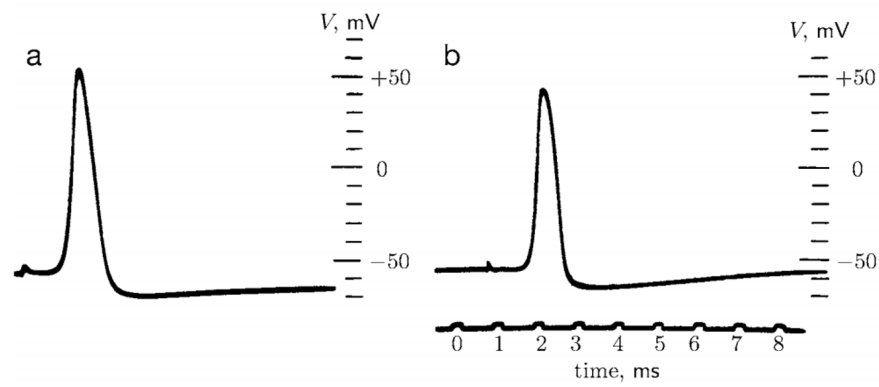
That is, when the channels were inserted and the  $K^+$  flowed out until equilibrium was reached, the internal  $K^+$  concentration went from  $100$  to  $99.99999$ .



- $Ca^{2+}$ : ???
- What does the  $Na^+ - K^+$  pump do in the action potential? Is it the same thing as the ion channel?
  - hyper-polarization.
  - no.
- 3 ways for  $Na^+$ ,  $K^+$  to pass the membrane
  - leaky
  - ion channel
  - ion exchanger/ion pump
- Why HH only used  $Na^+$  and  $K^+$ ?



(Figure 12.7 of *Biological Physics* )



left panel: an axon filled with potassium sulfate solution.

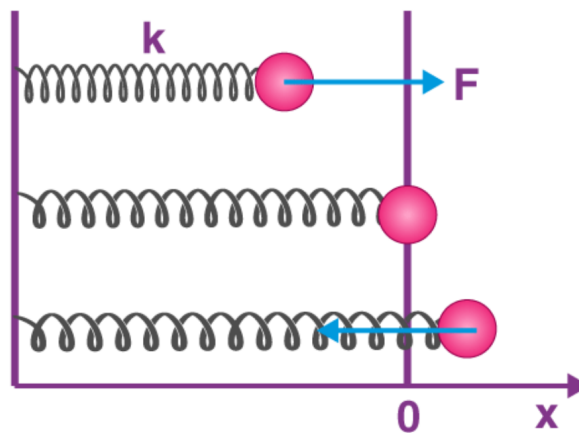
right panel: an normal axon.

(Figure 12.13 of *Biological Physics* )

# Nonlinear Dynamics in HH Model

## Introduction

Eg1: harmonic oscillator



Eg2: simple pendulum

$$\frac{dx}{dt} = \pm \sqrt{2a^2 \cos x - C_1}.$$

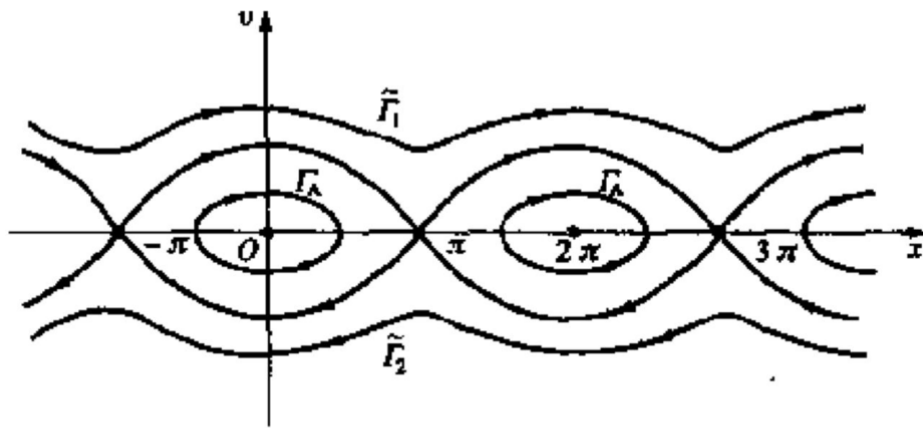


图 5-5

## Fixed Point

- nullcline
- vector field

## Limit Cycle

- Poincare-Bendixson TH

